

Clase 53 24 Noviembre 2020

Título de la nota

24/11/2020

$$PV = nRT$$

$P \rightarrow 0$ cambio de fase

$$V = \frac{nRT}{P}$$

$$T \rightarrow 0 \quad V \rightarrow 0$$
$$P \rightarrow \infty$$

Tercera Ley Termodinámica

$T = 0K$ cristal perfecto $S = 0$

Von der Waals (1876)



$$V_{\text{exclusión}} = \frac{4}{3} \pi d^3 \quad \text{para 2 moléculas}$$

$$\bar{V}_{\text{exclusión}} = \frac{2}{3} \pi d^3 \quad \text{por molécula}$$

$$\overline{V}_{\text{exclusión}} = \frac{2}{3} \tilde{n} d^3 N = b$$

$$b = \frac{L}{\text{mol}}$$

$$P = \frac{nRT}{V - nb}$$

$$\lim_{T \rightarrow 0} \boxed{V = nb}$$

$$P = \frac{RT}{\bar{v} - b}$$

$$\boxed{\begin{array}{l} V = nb = 0 \\ nb > V = - \\ \quad \quad \quad \times \end{array}}$$

$$F \propto C_1 C_2$$

$$C_1 = C_2 \text{ gas puro}$$

$$F \propto C^2$$

$$C = M = \frac{n}{V}$$

$$F \propto \left(\frac{n}{V}\right)^2$$

$$P = \frac{F}{A}$$

$$F = \frac{a' n^2}{V^2}$$

$$P = \frac{a' n^2}{V^2 A}$$

$$P = \frac{an^2}{V^2} = atm$$

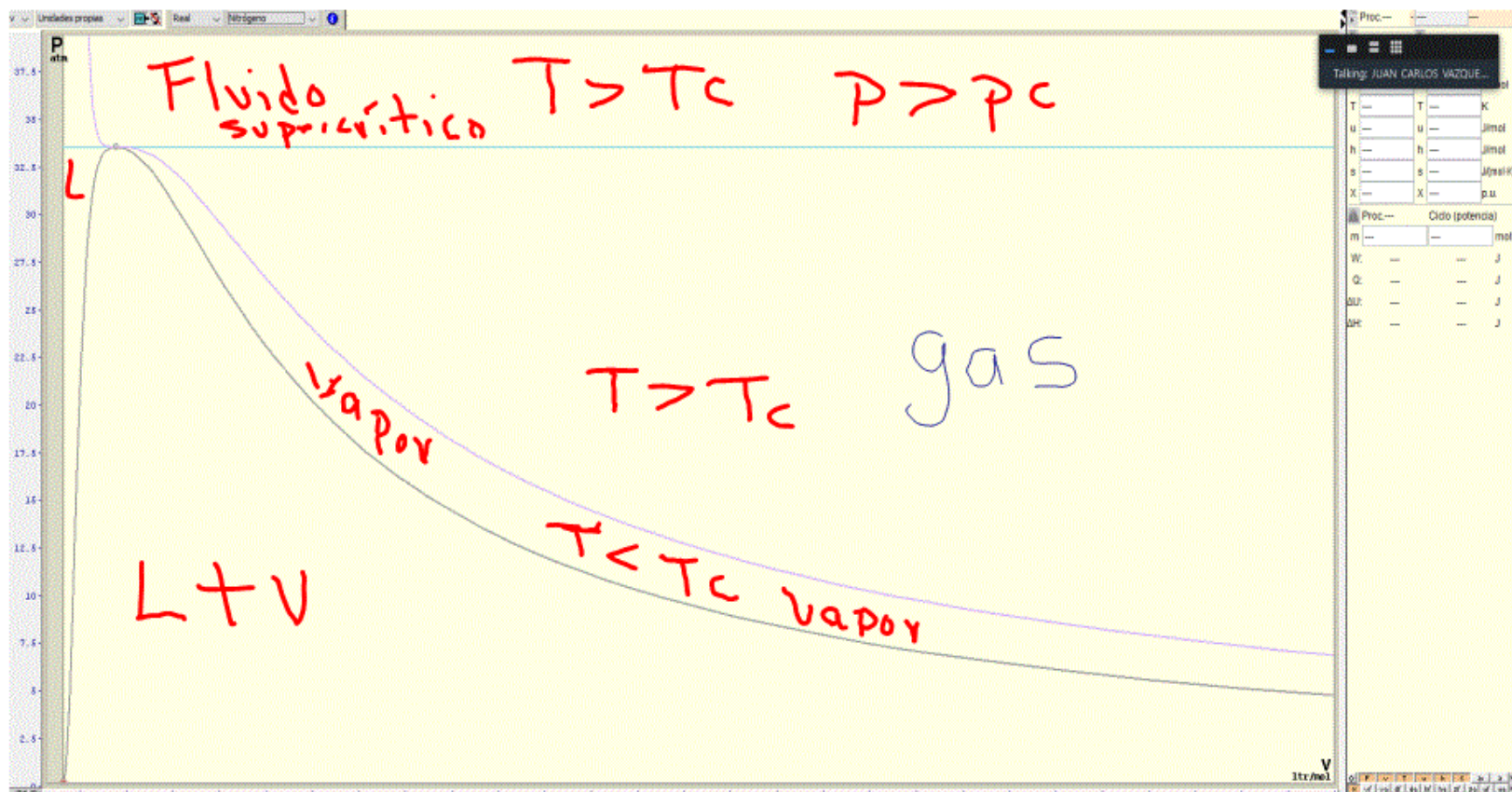
$$a = \frac{atm L^2}{mol^2}$$

$$P = \frac{nRT}{V-nb} - \frac{an^2}{V^2}$$

Extensivas
e intensivas

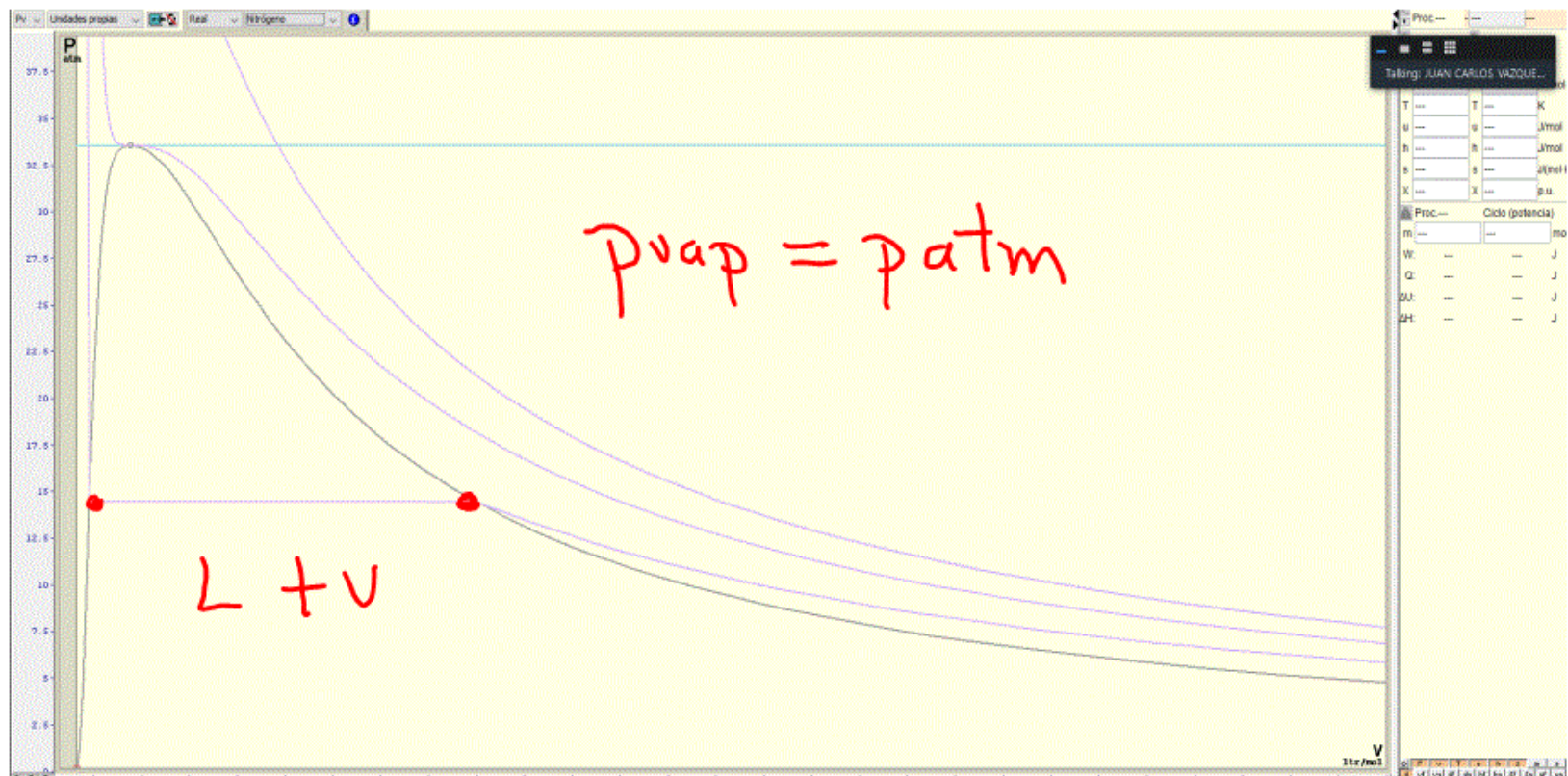
$$P = \frac{RT}{\bar{V}-b} - \frac{a}{\bar{V}^2}$$

intensivas

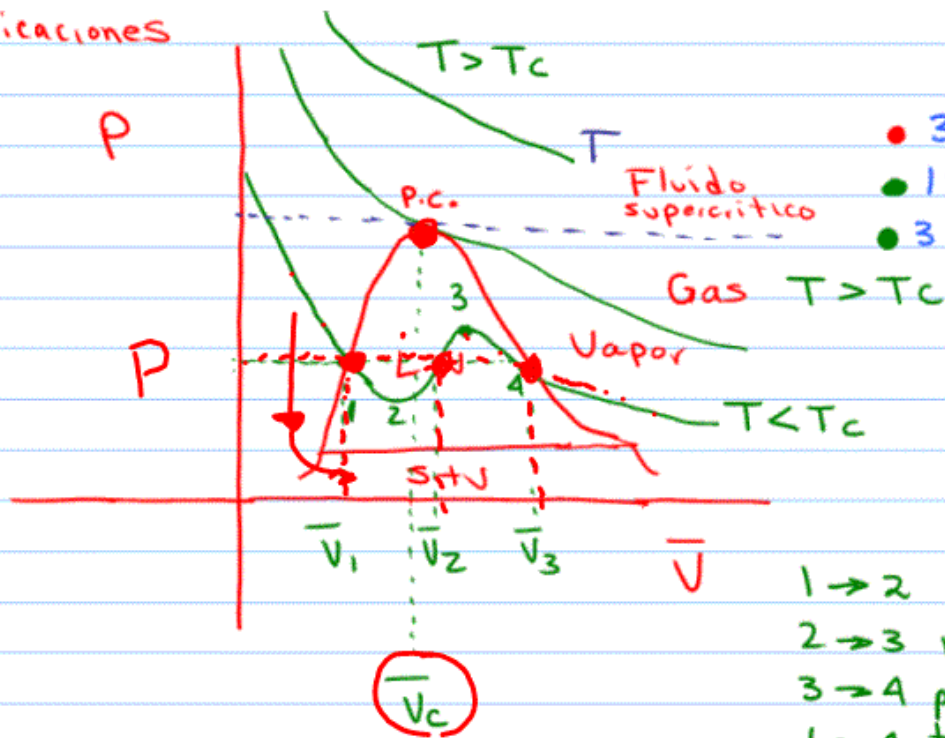


$$\text{vapor} \begin{cases} T < T_c \\ P < P_c \end{cases}$$

$$\text{gas} \begin{cases} T > T_c \\ P < P_c \end{cases}$$



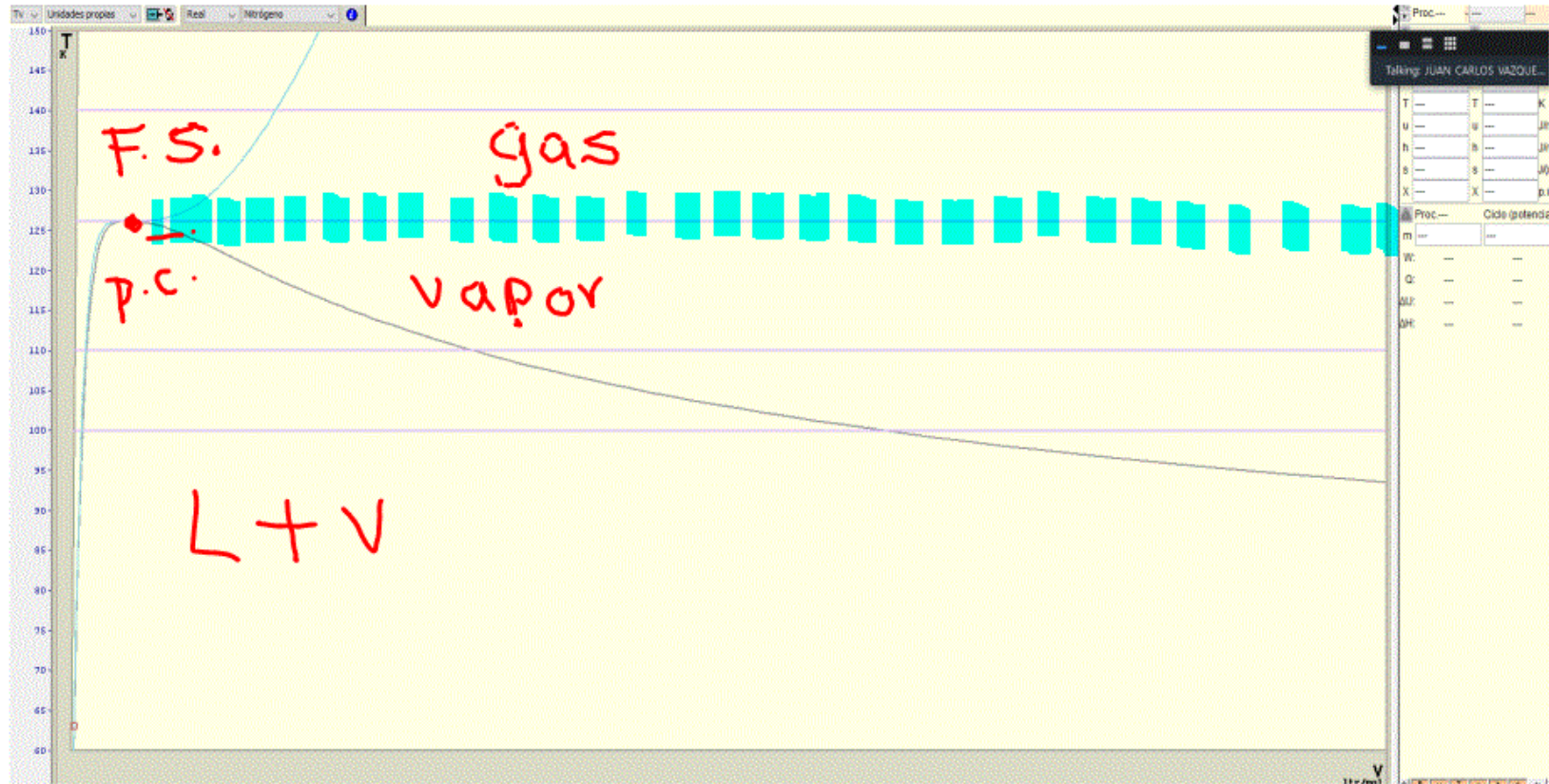
Implicaciones



esto implica que el volumen tiene:

- 3 raíces iguales reales
- 1 raíz real, 2 imaginarias
- 3 raíces reales \neq

- 1 \rightarrow 2 metaestable
- 2 \rightarrow 3 imposible
- 3 \rightarrow 4 posible vapor condensado
- 1 \rightarrow 4 transformación L \rightarrow V



$$\left(\frac{\partial T}{\partial \bar{v}} \right)_{p.c.} = 0$$

$$\left(\frac{\partial^2 T}{\partial \bar{v}^2} \right)_{p.c.} = 0$$

$$p = \frac{nRT}{V-nb} - \frac{an^2}{V^2}$$

$$p = \frac{RT}{\bar{v}-b} - \frac{a}{\bar{v}^2}$$

$$\left(P + \frac{a}{\bar{V}^2} \right) = \frac{RT}{\bar{V} - b}$$

$$\left(P + \frac{a}{\bar{V}^2} \right) (\bar{V} - b) = RT$$

$$P\bar{V} - Pb + \frac{a\bar{V}}{\bar{V}^2} - \frac{ab}{\bar{V}^2} = RT$$

$$T = \frac{P\bar{V}}{R} - \frac{Pb}{R} + \frac{a}{\bar{V}R} - \frac{ab}{\bar{V}^2 R}$$

$$T_c = \frac{p_c \bar{v}_c}{R} - \frac{p_c b}{R} + \frac{a}{\bar{v}_c R} - \frac{ab}{\bar{v}_c^2 R}$$

$$\left(\frac{\partial T}{\partial \bar{v}} \right)_{p,c} = \frac{p_c}{R} - \frac{a}{\bar{v}_c^2 R} + \frac{2\bar{v}_c ab}{\bar{v}_c^4 R}$$

$$= \frac{p_c}{R} - \frac{a}{\bar{v}_c^2 R} + \frac{2ab}{\bar{v}_c^3 R} = 0$$

$$\left(\frac{\partial T^2}{\partial \bar{v}^2} \right) = \frac{2a\bar{v}_c}{\bar{v}_c^4 R} - \frac{6\bar{v}_c^2 ab}{\bar{v}_c^6 R} = 0$$

$$= \frac{2a}{\bar{V}_c^3 R} - \frac{6ab}{\bar{V}_c^4 R} = 0$$

$$\left[= \frac{pc}{R} - \frac{a}{\bar{V}_c^2 R} + \frac{2ab}{\bar{V}_c^3 R} = 0 \right] \frac{3}{\bar{V}_c}$$

$$\frac{3pc}{\bar{V}_c R} - \frac{3a}{\bar{V}_c^3 R} + \frac{6ab}{\bar{V}_c^4 R}$$

$$\frac{3pc}{\bar{V}cR} - \frac{3a}{\bar{V}^3cR} + \frac{6ab}{\bar{V}^4cR} = 0$$

$$\frac{2a}{\bar{V}^3cR} - \frac{6ab}{\bar{V}^4cR} = 0$$

$$\frac{3pc}{\bar{V}cR} = \frac{a}{\bar{V}^3cR} = 0$$

$$\frac{3pc}{\bar{V}cR} = \frac{a}{\bar{V}^3cR}$$

$$a = 3pc \bar{V}^2$$

$\frac{\text{atm L}^2}{\text{mol}^2}$

$$\frac{2a}{\bar{V}_c^3 R} = \frac{6ab}{\bar{V}_c^4 R}$$

$$\frac{2(3pc\bar{V}_c^2)}{\bar{V}_c^3 R} = \frac{6(3pc\bar{V}_c^2)b}{\bar{V}_c^4 R}$$

$$\frac{\cancel{6pc}\bar{V}_c^{\cancel{2}}}{\cancel{\bar{V}_c^3}R} = \frac{\cancel{18pc}\bar{V}_c^{\cancel{2}}b}{\cancel{\bar{V}_c^4}R}$$

$$b = \frac{1}{3} \bar{V}_c = \frac{L}{\text{mol}}$$

obtener v

$$p = \frac{nRT}{v-nb} - \frac{an^2}{v^2}$$

$$\left(p + \frac{an^2}{v^2} \right) = \frac{nRT}{v-nb}$$

$$\frac{pv^2 + an^2}{v^2} = \frac{nRT}{v-nb}$$

$$(pv^2 + an^2)(v-nb) = v^2 nRT$$

$$(pv^2 + an^2)(v - nb) = v^2 nRT$$

$$pv^3 - pv^2 nb + an^2 v - an^3 b = v^2 nRT$$

$$pv^3 - pv^2 nb - v^2 nRT + an^2 v - an^3 b$$

$$\underline{pv^3 - v^2 (pnb + nRT) + v an^2 - an^3 b = 0}$$

$$v^3 - v^2 \left(nb + \frac{nRT}{P} \right) + \frac{v an^2}{P} - \frac{an^3 b}{P} = 0$$

$$L^3 - L^2 \left(\frac{\cancel{\text{mol}} L}{\cancel{\text{mol}}} + L \right) + L \frac{\cancel{\text{atm}} L^2 \cancel{\text{mol}}}{\cancel{\text{mol}} \cancel{\text{atm}}} - \frac{ }{P} = 0$$

$$L^3 - L^3 + L^3 - L^3 = 0$$

$$\frac{an^3b}{P} = \frac{\cancel{atm}L^2}{\cancel{mol^2}atm} \frac{mol^3L}{mol}$$

$$V^3 - V^2 \left(nb + \frac{nRT}{P} \right) + \frac{Van^2}{P} - \frac{an^3b}{P} = 0$$

$$P = \frac{RT}{\bar{V} - b} - \frac{a}{\bar{V}}$$